## Target Your **Reading**

## The Periodic Table and Physical Properties

**CHAPTER 7** 

Use this to focus on the main ideas as you read the chapter.

- **1. Before you read** the chapter, respond to the statements below on your worksheet or on a numbered sheet of paper.
  - Write an **A** if you **agree** with the statement.
  - Write a **D** if you **disagree** with the statement.
- **2. After you read** the chapter, look back to this page to see if you've changed your mind about any of the statements.
  - If any of your answers changed, explain why.
  - Change any false statements into true statements.
  - Use your revised statements as a study guide.

Before You Read A or D	Statement	After You Read A or D
	<b>1.</b> The elements are arranged on the periodic table according to their atomic numbers and mass numbers.	
	2. The elements in a group have similar properties.	
	<b>3</b> . Metals are located on the right side of the periodic table.	
	4. Not all isotopes are radioactive.	
	5. Radioactive elements have unstable nuclei.	
	<b>6</b> . An element's mass number is the number of neutrons in its nucleus.	
	7. Transmutation is another word for half-life.	
	8. Melting and boiling points change with pressure.	
	<b>9.</b> Thermal conductivity occurs because particles collide with one another.	
	<b>10</b> . Density is a physical property that depends on the size of a sample.	